

Application Memo

Concentration of 48%-Sodium Hydroxide

| | |
|--------------------|-----------------------------------|
| Industry | Inorganic chemical industry |
| Instrument | Automatic potentiometric titrator |
| Measurement method | Neutralization titration |
| Standards | JIS K1200-2 |

1. Overview

Sodium hydroxide concentration is specified by potentiometric titration with hydrochloric acid according to the JIS K 1200-2 Appendix 2.

Two endpoints appear in this titration and the target endpoint for sodium hydroxide is the first endpoint around pH7. This can be easily detected by setting the measurement condition that requires only one endpoint. The concentration of sodium hydroxide is calculated from the titration volume at the first endpoint.

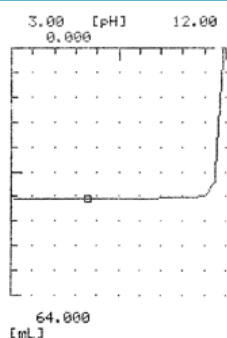
2. Apparatus

| | |
|-----------|--|
| Main unit | Automatic potentiometric titrator (preamplifier STD) |
| Electrode | Combined glass electrode (Internal solution 3.33M-potassium chloride) Temperature compensation electrode |

3. Reagents

| | |
|---------|--|
| Titrant | 1mol/L hydrochloric acid |
| Solvent | Water (Ion exchanged or distilled water) |

4. Example



—Titration curve—

—Measurement results—

| | Sample (g) | Titer (mL) | NaOH (%) |
|---------|---------------|---------------|-------------|
| 1 | 80.005 | 38.8742 | 48.833 |
| 2 | 80.005 | 38.8767 | 48.836 |
| 3 | 80.005 | 38.8813 | 48.842 |
| Average | | | 48.837 |
| SD | | | 0.005 |
| RSD(%) | | | 0.01 |

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