

Application Memo

Determination of Trivalent Chromium in Nickel Plating Solution

Industry	Non-ferrous metal
Instrument	Automatic potentiometric titrator
Measurement method	Oxidation reduction titration
Standards	

1. Overview

After adding 20% sodium hydroxide and hydrogen peroxide to the diluted sample, slowly boil it and cool. After adding pure water, ammonium fluoride, 3mol/L sulfuric acid and potassium iodide to it, trivalent chromium concentration is measured by titration with the 0.1mol/L sodium thiosulfate solution. The endpoint is the maximum inflexion on the titration curve. The trivalent chromium concentration is calculated from the titration volume of sodium thiosulfate.

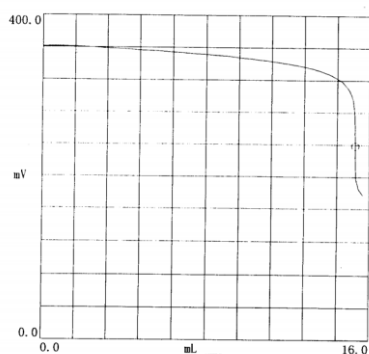
2. Apparatus

Main unit	Automatic potentiometric titrator (preamplifier STD)
Electrode	Platinum electrode Ceramic reference electrode

3. Reagents

Titrant	0.1mol/L sodium thiosulfate solution
Solvent	Pure water
Additive	20% sodium hydroxide, Hydrogen peroxide, 3mol/L sulfuric acid (6N) Ammonium fluoride, Potassium iodide

4. Example



-Titration curve-

-Measurement results-

	Concentration (g/L)
Average	2.6696
SD	0.3763
RSD(%)	14.10

Please feel free to contact us for any further information.

<Contact> Kyoto Electronics Manufacturing Co., Ltd.

Overseas Sales & Marketing Sect.

<http://www.kyoto-kem.com/en/contact/form.php>