

**Application Memo**

**Quantitative Determination of Hydrogen Peroxide and Peracetic Acid in Cleaning Solution**

Industry	Inorganic chemical industry
Instrument	Automatic potentiometric titrator
Measurement method	Precipitation titration
Standards	

**1. Overview**

After adding water and 25% sulfuric acid solution to the sample, it is titrated with 0.02mol/L potassium permanganate solution. The endpoint is the maximum inflexion on the titration curve. The hydrogen peroxide concentration is calculated from the titration volume of the titrant. After adding 20% potassium iodine solution to the titrated sample, it is titrated with 0.1mol/L sodium thiosulfate solution. The endpoint is the maximum inflexion on the titration curve. The peracetic acid concentration is calculated from the titration volume of the titrant.

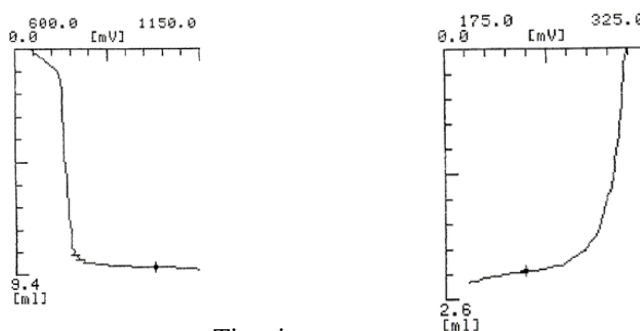
**2. Apparatus**

Main unit	Automatic potentiometric titrator (preamplifier STD)
Electrode	Platinum electrode, Ceramic reference electrode

**3. Reagents**

Titrant	0.02mol/L(0.1N) potassium permanganate solution 0.1mol/L sodium thiosulfate solution
Solvent	Water, 25% sulfuric acid solution, 20% potassium iodine solution

**4. Example**



—Titration curve—  
—Measurement results—

	Hydrogen Peroxide			Peracetic Acid		
	Sample(g)	Titer(mL)	Conc.(%)	Sample(g)	Titer(mL)	Conc.(%)
1	5.0	9.1253	0.3103	5.0	2.3197	2.320
2	5.0	9.2012	0.3128	5.0	2.2869	2.287
3	5.0	9.1382	0.3107	5.0	2.2662	2.266
Average			0.3113			2.291
SD			0.0138			0.027
RSD(%)			0.443			1.2

Please feel free to contact us for any further information.

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