KYOTO ELECTRONICS MANUFACTURING CO., LTD.

TIE-92004enL

Application Memo Chloride in Soap

| Industry | Cosmetics soap |
|---------------------------------|-----------------------------------|
| Instrument | Automatic potentiometric titrator |
| Measurement method Standards | Precipitation titration |

1. Overview

The chloride (converted to NaCl) in soap is measured by titration with the 0.1ml/L silver nitrate solution after methanol is added to the test sample. The endpoint is determined by the inflexion point on the titration curve. Chloride concentration is calculated from the titration volume of silver nitrate.

2. Apparatus

| Main unit | Automatic potentiometric titrator (preamplifier STD) |
|-----------|--|
| Electrode | Silver electrode |
| | Mercury sulfate reference electrode |

3. Reagents

| Titrant | 0.1mol/L silver nitrate solution |
|---------|----------------------------------|
| Solvent | Methanol |

4. Example

| 0.0 [mV] | 350.0 |
|-------------|-------|
| | 77 |
| - | 1 |
| - | 1 |
| | / |
| | |
| | |
| [] | |
| L | |
| 4.0 (m1) | |
| Titration | |

| | —Measure | ement results- | _ |
|---|--------------|----------------|------------------------|
| | Sample Lifer | | Chloride concentration |
| | (mL) | (mL) | (%) |
| 1 | 3.2139 | 2.8540 | 0.52171 |
| | | | |

—Titration curve—

Please feel free to contact us for any further information.

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