

TIQ-99327enL

Application Memo Purity of Amidosulfuric Acid

Industry Inorganic chemical industry
Instrument Automatic potentiometric titrator

Measurement method Acid-base titration Standards JIS K 8587

1. Overview

According to the standard testing method JIS K 8587-2018 Amidosulfuric acid (Reagent), amidosulfuric acid is measured by titration with 0.1mol/L sodium hydroxide after the sample is diluted with water.

The endpoint is the maximum inflexion on the titration curve. The purity of amidosulfuric acid is calculated from the titration volume of sodium hydroxide.

 $HOSO_2NH_2 + NaOH \rightarrow NaOSO_2NH_2 + H_2O$

2. Apparatus

Main unit Automatic potentiometric titrator (preamplifier: STD)

Electrode Combined glass electrode

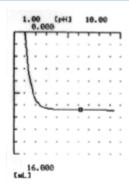
Temperature compensation electrode

3. Reagents

Titrant 0.1 mo1/L sodium hydroxide

Solvent Pure water

4. Example



—Measurement results—			
	Sample	Titer	Amidosulfuric
	- Constant		acid
	(g)	(mL)	(%)
1	4.8	10.1814	98.13
2	4.8	10.1836	98.15
3	4.8	10.1945	98.26
Average			98.18
SD			0.07
RSD(%)			0.07

[—]Titration curve—

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