

TIQ-99335enL

Application Memo

Concentration of Formic Acid

Industry Inorganic chemical industry
Instrument Automatic potentiometric titrator

Measurement method Acid-base titration

Standards

1. Overview

Formic acid is measured by titration with 1mol/L sodium hydroxide after the sample is diluted with water.

The endpoint is the maximum inflexion on the titration curve. The concentration of formic acid is calculated from the titration volume of sodium hydroxide.

$$HCOOH + NaOH \rightarrow HCOONa + H_2O$$

2. Apparatus

Main unit Automatic potentiometric titrator (preamplifier STD)

Electrode Combined glass electrode

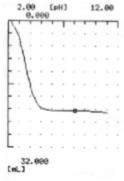
Temperature compensation electrode

3. Reagents

Titrant 1 mo1/L sodium hydroxide

Solvent Pure water

4. Example



—Measurement results—			
	Sample	Titer	Formic acid
	(g)	(mL)	(%)
1	1.0357	22.7929	99.106
2	1.0884	23.9522	99.115
3	1.1049	24.2749	98.953
Average			99.058
SD			0.091
RSD(%)			0.091

—Titration curve—

Please feel free to contact us for any further information.

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