

TIZ-99204enL

## **Application Memo**

# **Measurement of Low Concentration Sodium Chloride Solution**

Industry Inorganic chemical industry
Instrument Automatic potentiometric titrator

Measurement method Precipitation titration

Standards

# 1. Overview

Using the prepared 1 w/w% NaCl as a sample, titration was performed with  $0.01 \, \text{mol/L}$  silver nitrate, and the maximum inflection point in potentiometric titration was determined as the end point. Then, the concentration was determined by converting the titer of  $0.01 \, \text{mol/L}$  silver nitrate up to the end point into salt content.

#### 2. Apparatus

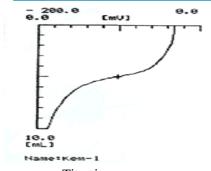
Main unit Automatic potentiometric titrator

Electrode Combined silver electrode

# 3. Reagents

Titrant 0.01 mol/L silver nitrate

## 4. Example



—Measurement results—			
	Sample size	Titer	Concentration
	(g)	(mL)	(w/w%)
1	0.2916	5.0945	1.019
2	0.2883	5.0095	1.013
3	0.2930	5.0775	1.011
Average			1.014
SD			0.004
RSD(%)			0.4

—Titration curve—

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